Effect Of Quenching Temperature On Structure And Mechanical Properties Of Aluminum Alloy With High Magnesium Content Al-10% Mg

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Abstract: Development of new high element aluminium alloys requires of solving features related to the modes of their thermal hardening. In the present work the quenching temperature interval choice of alloy with a high content of magnesium Al-10% Mg is determined. For this purpose, water quenching of extruded bars from alloy was carried from the temperatures leying in the region of the maximum solubility of magnesium in aluminum. The change of Brinell, Rockwell hardness, tensile strength, yield strength, elongation, toughness of alloy samples, cut at an angle of 0 and 90 ° to the axis of compression were found. The analysis of the microstructure using optical metallographic microscope at a magnification of 500 was worked out. It is shown that the tensile strength, elongation, toughness, hardness decreases, yield stress of the samples cut in the longitudinal and transverse direction rise with increasing quenching temperature. The temperature of quenching is set, providing a combination of high strength and toughness properties. The results obtained are discussed from the perspective of the completeness of dissolution of dispersed intermetallic phases. Results can be used during editing operations and wrought and semi-finished products stamping of aluminum alloy with a high content of magnesium.

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1. Introduction

Alloys of Al-Mg system are materials with one of the highest rates of specific strength. Magnesium additives help reduce the specific gravity, to increase strength while maintaining good corrosion resistance and weld ability (Mondolfo L. F., 1976).

Typically, the maximum content of magnesium in the wrought aluminum alloys is limited to 5-6%, for example alloys 5056, 5356 (Heat Treater's Guide, 1996). Rising of magnesium content above these values may causes in casting difficulties, hot and cold deformation and heat treatment (Fakhraei O. et al. 2013, Fakhraei O. et al, 2014). Modification using, applying of pulsed electromagnetic fields, other secondary treatment methods can solve a number of issues and improve casting formability of aluminum alloys containing about 10% magnesium (Fernández R. et al, 2012, Fourmeau M. et al, 2013, Zhivoderov V.M. et al, 1997). For example, the acoustic exposure causes an increase in the crystallization process of melt viscosity and heat and mass transfer coefficient variation due to changes in the physical properties of the melt. Furthermore, this effect decreases in the oxide and non-metallic inclusions content in the alloy. This is due to the ejection of non-metallic and oxide inclusions to the surface of the melt during product solidification. This increases the rate of heat transfer, launch new crystallization centers and the effect of structure refinement is observed (Aluminum Alloy Castings, 2004, MA Cheng-guo et al, 2014).

The maximum content of magnesium in refractor aluminum alloys is limited to 20% at nowdays, but this experimental alloys for shaped castings (Fakhraei O. et al, 2013), they are not serial and industrial. However, cast structure has several drawbacks: the heterogeneity of the dendrite structure, the presence of shrinkage and gas pores, element segregation. Ingot homogenization is applied as a first heat treatment in the technological process. Ingot with inhomogeneous thermodynamic unstable structure undergoes to annealing, which resulted in more homogeneous structure, ductility increases, which allows subsequent processing to greatly intensify pressure (compression) and reduce process waste. In many cases homogenization can improve the properties of deformable semi finished (Bensaada S. et al, 2011).

A significant portion of the casting defect can be eliminated by hot and cold working (Jie J.C. et al, 2013, Man-ping Liu et al, 2012). In the process of homogenized ingot hot rolling there heat cracks are formed caused by the presence of brittle β -phase, both separately and in a non-equilibrium eutectic (Sauvage X. et al, 2014).

Application of hydrostatic compression scheme lets receive in extruded panels' thickness of about 6 mm when pressing ingots in return for their hot rolling (Cepeda-Jiménez C.M. et al, 2009, Demirci H.i. et al, 2012). Further cold rolling requires addressing of ductility improve by softening heat treatment, including interoperation annealing. However, the mechanical properties set are provided by quenching and aging on the final stage of heat treatment (Golovin I.S. et al, 2013). The effectiveness of the hardening heat treatment of aluminum alloys depends on the aging mode and a homogeneous structure of a supersaturated solid solution after the quenching (Maisonnette D. et al, 2010). A presence of heterogeneous inclusions is possible at a reduced temperature of heating for quenching and slow cooling, which may prevent further decay in aging and reduce the set of properties (Zou B. et al, 2014, Grechnikov F.V. et al, 2009). Inflated heating temperature give rise to excessive thermal stresses lead to warping, raise the cost of production.

The goal of this work was to study the effect of quenching temperature on the structure, mechanical properties, the nature of hardening of aluminium alloy Al-10% Mg and the set of processing modes, providing a combination of high strength and plastic properties.

2. Materials and Metods

The chemical composition of the alloy was determined by energy-dispersive consoles of SEM Tescan Vega, presented in Table 1.

Table 1. Mass fraction of elements in the alloy Al-10%Mg.

. Mass fraction of elements, %							
Mg	Zr	Be	Ti	В	Cu	Со	Al
10,5	0,11	0,08	0,018	0,01	0,001	0,015	89,266

Manufacturing process of extruded semi-finished included:

- round ingots casting with diameter of 127 mm and a length of 700 mm in the semi-continuous casting installation in the mold DC solenoid (Zuo Y. et al, 2005);

- homogenizing the ingot at a temperature of 750K within 15-17 hours;

- hot-pressing the strip ($6 \times 15 \times 115$ mm) at a temperature of 700-720K with the air cooling.

From pressed strips the samples of rectangular cross-section of 6x10 mm in length 115 mm were cut in the direction 0 and 90° to the extrusion axis. The samples were heated to a temperature of 650 K, 690 K, 710 K, 740 K in the laboratory furnace, then cooling was in usual water. Quenching temperature is chosen so as to ensure of various dissolution completeness of disperse phase beta β (Al₈Mg₅). After quenching, the samples were tested for hardness Brinell HB (indenter diameter 2.5 mm), Rockwell HRB (indenter diameter 1,588 mm). Uniaxial tensile test was performed on the test vehicle Testometric. They are allowed to determine the tensile strength $\sigma_{\rm h}$ (MIIa), yield strength $\sigma_{0.2}$ (MIIa), elongation δ (%). Toughness KCU was evaluated by mechanical impact tester.

Since extruded semi-finished can undergo the cold deformation when straightening, bending,

hooking and other shaping operations, as well as work hardening to improve the strength characteristics the effect of material hardening at different modes of quenching was studied in the work.

To this aim, the curves of dependence the engineering strength and engineering deformation were obtained from tensile diagrams. For this purpose portion of the residual strain were divided into 5-6 parts (Figure 1). That gave the value of intermediate lengths $l_i=l_0+\Delta l_i$, mm. here l_0 – initial sample length.

For each intermediate length P_i was determined on the load axis, and then the current cross-sectional area was calculated $F_i = F_0 \times l_0 / l_i$, F_0 – initial sectional area of the sample, mm². Then the engineering strength was found by the formula: $\sigma_i = P_i / F_i$. Current deformation ε_i was calculated by formula: $\varepsilon_i = \ln(l_i / l_0) \times 100\%$.



Figure 1 - Diagram to find the current lengths and load

Microsections were produced for metallographic analysis of the thermally treated samples. The following composition was subjected to etching 95% H_2O , 2% HF, 3% HNO₃.

3. Results

Results of hardness depending upon the quenching temperature are shown in Figure 2.

From Figure 2 it cam be seen that hardnes drops due to rosing of quencing temperature. However hardnes HRb starts to rise slowly from temperature 655K.



Figure 2 - The hardness of samples depending upon the quenching temperature



Figure 3 - Change of tensile strength and yield strength depending on quenching temperature: σ_b^{\parallel} , $\sigma_{0,2}^{\parallel}$ – tensile strength and yield strength of the samples cut along the extrusion axis, σ_b^{\perp} , $\sigma_{0,2}^{\perp}$ – tensile strength and yield strength of the samples cut across the extrusion axis.

Figure 3 shows the growth in tensile strength σ_b (sigma b), decrease of yield strength $\sigma_{0,2}$ while temperature quenching rises. Figure 4 shows the dependence of the relative elongation and toughness upon the quenching temperature.



Figure 4 - The change in elongation and toughness as a function of the quenching temperature

Details of ductility and toughness increase with temperature quenching.

Figure 5 shows the curves of "engineering strength σ_i -engineering deformation ϵ_i " of the alloy samples quenched from different temperatures.



Figure 5 - Change the engineering strength from engineering deformation of alloy quenched from different temperatures.

As seen from these graphs, increasing the quenching temperature has an ambiguous effect on the character of the alloy hardening: the increase in the heating temperature from 650 to 690 K leads to a decrease in indicators of engineering strength, further increasing in temperature slightly increases and does not affect the nature of the curves.

Figure 6 shows the microstructure of the specimens in an equilibrium state, and after quenching from different temperatures.

As it follows from the microstructure in the equilibrium state in the alloy structure a substantial amount of dispersed phase is present. This is mainly β -phase, but large inclusions of magnesium silicide Mg₂Si also meet. The grain size is not equal, lays in interval 20-70 micrins and does not desapeared in process of heating.



Figure 6 - The microstructure of the samples, x500: a - in the annealed condition, b, c, d - after quenching temperature 653K, 693K and 713K respectively.

4. Discussion

These results show that quenching efficiency increases sharply at temperatures above the 690K, while the high strength and plastic characteristics are stored before hardening temperatures of 740K. Solvus temperature of the alloy according to (Mondolfo L. F., 1976), is at 650K. At temperatures above the solvus line 40 ... 50 K (β-phase Al₈Mg₅ and other intermetallic phases is like Al₃Zr) almost completely transformed into α -solid solution. Often formula β (beta) - phase is as Al₃Mg₂ (Golovin I.S. et al, 2013), but it has a magnesium content of about 37.5% by weight, whereas in the phase of Al₈Mg₅ composition contains about 36 wt.% Magnesium that on Al-Mg diagram is located closer to the left and to the aluminum corner. Therefore it is more likely that in equilibrium with the solid solution is phase Al₈Mg₅.

Solidus temperature which bounds the area of a solid solution from top limit is at 800K. Heating to these temperatures can lead to a burnout of the structure and melting of the grain boundaries. Fixing by quenching the homogeneous solid solution is accompanied by a sharp increase of enough tensile strength and especially plastic characteristics - elongation and toughness (Avazkonandeh-Gharavol M.H. et al, 2014).

Quenching of the longitudinal and transverse specimens accompanied identical nature of the change properties. At temperatures above 690K quenching difference in properties is reduced and does not exceed 2%. The dispersion microstructure formed during crystallization of the alloy is grains of the solid solution of aluminum primary and secondary intermetallic compounds, such as Zn₂Al₃, TiAl₃, Al₈Mg₅, which provide improved processability of alloy (Liu Feng-xiang et al, 2009).

It is nessesary to note that uneven grain size, which is usual for hot extruded profiles, remains in microstructure of samples quenched from all choosen temperatures. As rule, the unevent grain leads to a low stampability of the flat workpieces. So in this case it's effect didn't observe.

Thus, the optimal in terms of a combination of high strength and ductility for extruded products made of alloy with a high content of magnesium is quenching from heat temperature 710 ... 740 K and water cooling with temperature 320...330 K. Mechanical properties after this mode of heat treatment are tensile strength - 410-420 Mpa, yield strength 185-200 MPa, elongation of 30-32%, the toughness of 1.15 MJ/m2, which exceeds similar characteristics of the alloy 5056 and is located at level of the low-carbon steel type A620, A622. The alloy of Al-10% Mg has high hardenability at cold deformation. This makes it possible to expose not only the semi-hardened tensile straightening but also significantly deform them in a cold state increase by providing strength characteristics of tensile strength to 530-550 MPa at a sufficient level of ductility (5.6% elongation). (21, 8, 15).

5. Conclusion

Studies have helped to identify the optimal mode of hardening high-magnesium alloy that keeps the optimal ratio of strength and ductility characteristics.

The microstructure of samples heat treated at these modes confirmed the physical picture of the structural transformations.

The results confirmed the possibility of thermal hardening aluminum alloy with high content of magnesium by quenching. Future studies related to the study of properties and structural transformations at the subsequent natural and artificial aging.

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References

- 1. Aluminum Alloy Castings: Properties, Processes, and Applications by J. G. Kaufman. Published by AFS and ASM International in 2004. ISBN 0871708035.
- Avazkonandeh-Gharavol M.H., 2014, M. Haddad-Sabzevar, H. Fredriksson. On the Microsegregation of Al-Mg Alloys by Thermal Analysis and Numerical Modeling. Journal of Alloys and Compounds, In Press, Accepted Manuscript, Available online 15 May 2014. DOI 10.1016/j.jallcom.2014.05.026.
- Bensaada S., 2011, M.T. Bouziane, F. Mohammedi. Effect of the temperature on the mechanism of the precipitation in Al-8% mass. Mg alloy. Materials Letters, Volume 65, Issues 17 - 18, September 2011, Pages 2829-2832. DOI:10.1016/j.matlet.2011.05.087.
- 4. Cepeda-Jiménez C.M., 2009, M. Pozuelo, O.A. Ruano, F. Carreño. Influence of the thermal treatment on the microstructure and hardness

evolution of 7075 aluminium layers in a hot-rolled multilayer laminate composite. Journal of Alloys and Compounds 478 (2009) 154–162. DOI 10.1016/j.jallcom.2008.11.078.

- Demirci H.i., 2012, H. Evlen Effect of extrusion ratio on the wear behavior of Al – Si and Al – Mg alloys. Review Article. Journal of Alloys and Compounds, Volume 510, Issue 1,5 January 2012, Pages 26-32 DOI 10.1016/j.jallcom.2011.08.074
- Fakhraei O., 2014, M. Emamy. Effects of Zr and B on the structure and tensile properties of Al - 20% Mg alloy. Materials & Design, Volume 56, April 2014, Pages 557-564 DOI 10.1016/j.matdes.2013.11.083
- Fakhraei O., 2013, M. Emamy, H. Farhangi The effect of Al - 5Ti - 1B grain refiner on the structure and tensile properties of Al - 20% Mg alloy. Materials Science and Engineering: A, Volume 560, 10 January 2013, Pages 148-153 DOI 10.1016/j.msea.2012.09.050
- Fernández R., 2012, G. González-Doncel A unified description of solid solution creep strengthening in Al - Mg alloys. Materials Science and Engineering: A, Volume 550, 30 July 2012, Pages 320-324 DOI 10.1016/j.msea.2012.04.080
- Fourmeau M., 2013, T. Børvik, A. Benallal, O.S. Hopperstad. Anisotropic failure modes of highstrength aluminium alloy under various stress states. International Journal of Plasticity, Volume 48, September 2013, Pages 34-53. DOI 10.1016/j.ijplas.2013.02.004
- Golovin I.S., 2013, A.V. Mikhaylovskaya, H.-R. Sinning. Role of the β-phase in grain boundary and dislocation anelasticity in binary Al – Mg alloys. Journal of Alloys and Compounds, Volume 577, 15 November 2013, Pages 622-632. DOI 10.1016/j.jallcom.2013.06.138.
- Liu Feng-xiang, 2009, Rang-su Liu, Zhao-yang Hou, Hai-Rong Liu, Ze-an Tian, Li-li Zhou. Formation mechanism of atomic cluster structures in Al - Mg alloy during rapid solidification processes. Annals of Physics, Volume 324, Issue 2, February 2009, Pages 332-342. DOI: 10.1016/j.aop.2008.10.010.
- Grechnikov F.V., 2009 Noskova E.A., Savel'eva O.G. A Study of the anisotropy of properties of sheet semiproducts from alloy AMg10 Metal Science and Heat Treatment. July 2009, Volume 51, Issue 7-8, pp 326-329. DOI: 10.1007/s11041-009-9163-3.
- Heat Treater's Guide: Practices and Procedures for Nonferrous Alloys edited by Harry Chandler. Published by ASM International, 1996. ISBN 0871705656, 669 pages.
- 14. Jie J.C., 2013, C.M. Zou, E. Brosh, H.W. Wang, Z.J. Wei, T.J. Li Microstructure and mechanical

properties of an Al – Mg alloy solidified under high pressures. Journal of Alloys and Compounds, Volume 578, 25 November 2013, Pages 394-404. DOI: 10.1016/j.jallcom.2013.04.184.

- Man-ping LIU, 2012, Shao-chun SUN, Hans J. ROVEN, Ying-da YU, Zhen ZHANG, Maxim MURASHKIN, Ruslan Z. VALIEV. Deformation defects and electron irradiation effect in nanostructured Al - Mg alloy processed by severe plastic deformation. Transactions of Nonferrous Metals Society of China, Volume 22, Issue 8, August 2012, Pages 1810-1816. DOI 10.1016/S1003-6326(11)61391-5.
- MA Cheng-guo, 2014, Shu-yan QI, Shuang LI, Huan-yan XU, Xiu-lan HE. Melting purification process and refining effect of 5083 Al – Mg alloy. Transactions of Nonferrous Metals Society of China, Volume 24, Issue 5, May 2014, Pages 1346-1351 DOI:10.1016/S1003-6326(14)63198-8.
- Maisonnette D., 2010, M. Suery, D. Nelias, P. Chaudet, T. Epicier Effects of heat treatments on the microstructure and mechanical properties of a 6061aluminium alloy Materials Science and Engineering: A, Volume 528, Issue 6, 15 March 2011, Pages 2718-2724 DOI 10.1016/j.msea.2010.12.011.
- Mondolfo L. F., 1976. Aluminum Alloys: Structure and Properties. Butterworth & Co (Publishers) Ltd, 1976. ISBN: 978-0-408-70932-3, 971 pages.
- Sauvage X., 2014, N. Enikeev, R. Valiev, Y. Nasedkina, M. Murashkin. Atomic-scale analysis of the segregation and precipitation mechanisms in a severely deformed Al Mg alloy. Acta Materialia, Volume 72, 15 June 2014, Pages 125-136 DOI 10.1016/j.actamat.2014.03.033.
- 20. Zhivoderov V.M., 1997 Bibikov A.M., Inozemtzev A.L. A method of casting aluminum alloys, the aluminum alloy and a method for manufacturing intermediate products thereof. Patent for an invention RUS 2111826, 1997.
- Zou B., 2014, Z.Q. Chen, C.H. Liu, J.H. Chen Vacancy - Mg complexes and their evolution in early stages of aging of Al - Mg based alloys Original Research Article Applied Surface Science, Volume 298, 15 April 2014, Pages 50-55 DOI 10.1016/j.apsusc.2014.01.078.
- Zuo Y., 2005, J. Cui, J. Dong, F. Yu. Effect of low frequency electromagnetic field on the constituents of a new super high strength aluminum alloy. Journal of Alloys and Compounds, Volume 402, Issues 1 2, 27 October 2005, Pages 149 155. DOI:10.1016/j.jallcom.2005.04.135.

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