

Novel Quinazoline Derivatives Bearing A Sulfonamide Moiety As Anticancer and Radiosensitizing Agents

Mostafa M. Ghorab^{1,2*}; Fatma A. Ragab³; Helmi I. Heiba² and Ahmad A. Bayomi²

¹Department of Pharmacognosy, College of Pharmacy, King Saud University, P.O. Box 2457, Riyadh 11451, Saudi Arabia.

²Department of Drug Radiation Research, National Centre for Radiation Research & Technology, Atomic Energy Authority, Cairo, Egypt.

³Department of Pharmaceutical Chemistry, Faculty of Pharmacy, Cairo University, Cairo, Egypt.
mmsghorab@yahoo.com

Abstract: Quinazoline derivatives possess many types of biological activities and have recently been reported to show substantial antitumor activity *in vitro* and/or *in vivo*. There is a variety of mechanisms for their anticancer activity. The present work reports the possible utility of methyl anthranilate in the synthesis of some new quinazoline derivatives, bearing a substituted sulfonamide moiety. All the newly synthesized compounds were evaluated for their *in vitro* anticancer activity against human liver cancer cell line (HEPG2), using doxorubicin as a reference drug. In addition, the most active compounds **14** and **15** were selected and evaluated for their ability to enhance the cell killing effect of γ -radiation.

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1. Introduction

Quinazolines are an important class of heterocyclic compounds, which represent building block for approximately 150 naturally occurring alkaloids [1,2] and have relieved an increasing interest from medicinal chemists due to their wide range of biological and pharmaceutical activities including antidiabetic [3] antihyperglycemic [4], antihypertensive [5], antihistaminic [6], antioxidant [7], anti-inflammatory [8], antipsychotic, muscle relaxant [9], anticonvulsant [10], antiparkinsonian [11], antitoxoplasmic [12], antitubercular [13], antibacterial [14], antifungal [15] and antiviral activities [16]. Recently, quinazoline derivatives have been reported to show substantial antitumor activity *in vitro* and/or *in vivo* [17-21]. From the literature survey it was found that quinazoline derivatives act as antitumor agents through a variety of mechanisms such as dihydrofolate reductase inhibition [22], microtubule Polymerization inhibition [23], epidermal growth factor receptor inhibition [24], cyclin-dependent kinase inhibition [25], thymidylate synthase inhibition [26], selective erbB2 receptor tyrosine kinase inhibition [27], selective c-Src kinase inhibition [28] and tumor suppressor protein (p-53) reactivator [29]. According to recent data, quinazoline derivatives show inhibition of the growth of human hepatocellular cancer cells [30,31] and have synergistic effect when combined with γ -radiation through potentiation of the antitumor effect of single and multiple fractions of γ -radiation [32,33]. Sulfonamides constitute an important class of drugs,

with several types of pharmacological activities including antibacterial [34], anti-carbonic anhydrase [35], diuretic [36], hypoglycemic [37] and antithyroid activity [38]. Also, some structurally novel sulfonamide derivatives have recently been reported to show substantial antitumor activity *in vitro* and/or *in vivo* [39-42]. In the light of these facts, we planned to synthesize a novel quinazoline derivatives, in order to study their structure activity relationship and hoping that the new compounds might show significant anticancer activity. Moreover, we also aimed to evaluate these new compounds for their *in vitro* anticancer activity in combination with γ -radiation, to evaluate their ability to enhance the cytotoxic activity of γ -radiation.

2. Experimental

2.1. Chemistry

Melting points are uncorrected and were determined on a Stuart melting point apparatus (Stuart Scientific, Redhill, UK). Elemental analysis (C, H, N) were performed on Perkin-Elmer 2400 analyser (Perkin-Elmer, Norwalk, CT, USA) at the microanalytical laboratories of the Faculty of Science, Cairo University. All compounds were within $\pm 0.4\%$ of the theoretical values. The IR spectra (KBr) were measured on Shimadzu IR 110 spectrophotometer (Shimadzu, Koyoto, Japan), ¹H-NMR spectra were obtained on a Bruker proton NMR-Avance 300 (300 MHz) (Bruker, Munich, Germany), in DMSO-d₆ as a solvent, using tetramethylsilane (TMS) as internal standard. Mass

spectra were run on HP Model MS-5988 (Hewlett Packard, Palo, Alto, California, USA). All reactions were monitored by thin layer chromatograph (TLC) using precoated Aluminium sheets Silica gel Merck 60 F254 and were visualized by UV lamp (Merck, Darmstadt, Germany).

2.1 Methyl 2-isothiocyanatobenzoate (2)

Prepared according to the previously reported procedure [43].

2.2 Methyl 2-(3-(4-(*N*-pyridin-2-ylsulfamoyl)phenyl)thioureido)benzoate (4)

A mixture of methyl-2-isothiocyanatobenzoate **2** (1.9 g, 0.01 mole) and pyridin-2-yl benzenesulfonamide **3** (2.49 g, 0.01 mol) in dimethylformamide (20 mL) was stirred at room temperature for 5 h. The reaction mixture was poured onto cold water, the obtained solid was crystallized from dioxane to give compound **4**: Yield, 94 %; m.p. 235-237 °C; IR (KBr, cm⁻¹): 3300, 3238 (NH), 3041 (CH arom.), 1651 (C=O), 1262 (C=S), 1389, 1133 (SO₂). ¹H-NMR (DMSO-d₆)δ: 3.6 [s, 3H, OCH₃], 7.0-8.0 [m, 14H, Ar-H+2NH], 10.26 [s, 1H, SO₂NH]. MS, *m/z* (%): 429 [M⁺ -CH₃] (0.4), 226 (100). Anal. Calcd. For C₂₀H₁₈N₄O₄S₂: C, 54.28; H, 4.10; N, 12.66. Found: C, 54.44; H, 3.99; N, 12.87.

2.3 4-(3-Amino-4-oxo-3,4-dihydroquinazolin-2-ylamino)-*N*-(pyridin-2-yl) benzenesulfonamide (5)

A mixture of compound **4** (4.4 g, 0.01 mol) and hydrazine hydrate (1 g, 0.02 mol) in ethanol (30 mL) was refluxed till evolution of hydrogen sulfide gas was ceased (lead acetate paper) (14 h.), The reaction mixture was filtered while hot and the solid obtained was crystallized from dioxane to give compound **5**: Yield, 80%; m.p. 255-257 °C; IR (KBr, cm⁻¹): 3410, 3308, 3183 (NH, NH₂), 3045 (CH arom.), 1660 (C=O), 1383, 1133 (SO₂). ¹H-NMR (DMSO-d₆)δ: 6.8 [s, 2H, NH₂, D₂O-exchangeable], 7.3-8.0 [m, 13H, Ar-H+NH], 13.2 [s, 1H, SO₂NH]. MS, *m/z* (%): 408 [M⁺] (2.1), 343 (100). Anal. Calcd. For C₁₉H₁₆N₆O₃S: C, 55.87; H, 3.95; N, 20.58. Found: C, 55.97; H, 3.75; N, 20.88.

2. 4-(9-Oxo-[1,2,4]triazolo[5,1-*b*]quinazolin-3(9*H*)-yl)-*N*-(pyridin-2-yl) benzene-sulfonamide (6)

A solution of compound **5** (4.0 g, 0.01 mol) in Formic acid (20 mL) was refluxed for 5 h. The reaction mixture was cooled and then poured onto cold water, the obtained solid was crystallized from dioxane to give compound **6**: Yield, 71%; m.p. 281-283 °C; IR (KBr, cm⁻¹): 3227 (NH), 3045 (CH arom.), 1663 (C=O), 1395, 1139 (SO₂). ¹H-NMR (DMSO-d₆)δ: 6.9-7.8 [m, 12H, Ar-H], 13.1 [s, 1H, SO₂NH], 8.0 [s, 1H, CH triazole]. MS, *m/z* (%): 418 [M⁺] (1.65), 88(100). Anal. Calcd. For C₂₀H₁₄N₆O₃S: C, 57.41; H, 3.37; N, 20.08. Found: C, 57.17; H 3.15; N, 19.8.

2.5. *N*-(4-oxo-2-(4-(*N*-pyridin-2-ylsulfamoyl)phenylamino)quinazolin-3(4*H*)-yl)acetamide (7)

A solution of compound **5** (4.0 g, 0.01 mol) in acetic anhydride (20 mL) was refluxed For 10 minutes, The formed solid mass was collected and crystallized from ethanol to give compound **7**. Yield, 68%; m.p. 149-151 °C; IR (KBr, cm⁻¹): 3190 (NH), 3065 (CH arom.), 1721, 1662 (2 C=O), 1393, 1142 (SO₂). ¹H-NMR (DMSO-d₆)δ: 2.6 [s, 3H, COCH₃], 7.3-8.2 [m, 13H, Ar-H+NH], 11.7 [s, 1H, SO₂NH]. Anal. Calcd. For C₂₁H₁₈N₆O₄S: C, 55.99; H, 4.03; N, 18.66. Found: C, 55.97; H, 3.88; N, 18.88.

2.6. 2-(4-Methyl-9-oxo-[1,2,4]triazolo[5,1-*b*]quinazolin-3(9*H*)-yl)-*N*-(pyridin-2-yl)benzenesulfonamide (8)

A solution of compound **5** (4.0 g, 0.01 mol) in acetic anhydride (20 mL) was refluxed For 10 h. The formed solid mass was collected and crystallized from ethanol to give compound **8**: Yield, 66%; m.p. 296-298 °C; IR (KBr, cm⁻¹): 3272 (NH), 1686 (C=O), 1359, 1169 (SO₂). ¹H-NMR (DMSO-d₆)δ: 1.2 [s, 3H, CH₃], 7.2-8.2 [m, 12H, Ar-H], 13.4 [s, 1H, SO₂NH]. Anal. Calcd. For C₂₁H₁₆N₆O₃S: C, 58.32; H, 3.73; N, 19.43. Found: C, 58.44; H 3.78; N, 19.22.

2. 7 4-(2,10-Dioxo-2,3-dihydro-1*H*-[1,2,4]triazino[3,2-*b*]quinazolin-4(10*H*)-yl)-*N*-(pyridin-2-yl)benzenesulfonamide (9)

A mixture of compound **5** (4.0 g, 0.01 mol) and ethyl bromoacetate (1.69 g, 0.01 mol) and sodium ethoxide (0.68 g, 0.01 mol) in ethanol (30 mL) was refluxed for 8 h. The reaction mixture was filtered while hot and the solid obtained was crystallized from dioxane to give compound **9**: Yield, 78%; m.p. 182-184 °C; IR (KBr, cm⁻¹): 3484 (NH) 3065 (CH arom.), 2922, 2846 (CH aliph.), 1743, 1675 (2C=O), 1366, 1136 (SO₂). ¹H-NMR (DMSO-d₆)δ: 3.9 [s, 2H, CH₂CO], 7.2-7.9 [m, 12H, Ar-H], 8.2 [s, 1H, NH], 13.2 [s, 1H, SO₂NH]. Anal. Calcd. For C₂₁H₁₆N₆O₄S: C, 56.24; H, 3.60; N, 18.74. Found: C, 55.97; H 3.75; N, 18.55.

2. 8 2-Cyano-*N*-(4-oxo-2-(4-(*N*-pyridin-2-ylsulfamoyl)phenylamino)quinazolin-3(4*H*)-yl)acetamide (11)

A solution of compound **5** (4.0 g, 0.01 mol) and ethyl cyanoacetate (10 mL) was refluxed for 8h. The formed solid mass was collected and crystallized from methanol to give compound **11**: Yield, 84 %; m.p. >300 °C; IR (KBr, cm⁻¹): 3223 (NH), 2934, 2867 (CH aliph.), 2208 (CN), 1652 (C=O), 1398, 1139 (SO₂). ¹H-NMR (DMSO-d₆)δ: 4.1 [s, 2H, CH₂CN], 7.4-8.0 [m, 14H, Ar-H+2NH], 13.1 [s, 1H, SO₂NH]. MS, *m/z* (%): 475 [M⁺] (0.15), 373 (100). Anal. Calcd. For C₂₂H₁₇N₇O₄S: C, 55.57; H, 3.60; N, 20.62. Found: C, 55.44; H 3.88; N, 20.42.

2.9 4-(2-(Cyanomethyl)-9-oxo-[1,2,4]triazolo[5,1-b]quinazolin-3(9H)-yl)-N-(pyridin-2-yl)benzenesulfonamide (12)

A mixture of compound **5** (4.0 g, 0.01 mol) and ethyl cyanoacetate (10 mL) in dimethylformamide (10 mL) containing 3 drops of triethylamine was refluxed for 8h. The formed solid mass was collected and crystallized from dioxane to give compound **12**: Yield, 84%; m.p. >300 °C; IR (KBr, cm⁻¹): 3356 (NH), 2928, 2870 (CH aliph.), 2206 (CN), 1645 (C=O), 1393, 1139 (SO₂). ¹H-NMR (DMSO-d₆)δ: 4.2 [s, 2H, CH₂CN], 7.3-8.0 [m, 12H, Ar-H], 13.4 [s, 1H, SO₂NH]. MS, m/z (%): 457 [M⁺] (18.8), 92 (100). Anal. Calcd. For C₂₂H₁₅N₇O₃S: C, 57.76; H, 3.30; N, 21.43. Found: C, 57.97; H, 3.65; N, 21.22.

2. 10 4-(2-Sulfanylidine-9-oxo-[1,2,4]triazolo[5,1-b]quinazolin-3(9H)-yl)-N-(pyridin-2-yl)benzenesulfonamide (13)

A solution of compound **5** (4.0 g, 0.01 mol) in carbondisulfide (1.52 g, 0.02 mol) in pyridine (20 mL) was refluxed for 10 h. The reaction mixture was cooled and then poured onto cold water, the obtained solid was crystallized from acetic acid to give compound **13**: Yield, 68 %; m.p. >300 °C; IR (KBr, cm⁻¹): 3223 (NH), 3038 (CH arom.), 1663 (C=O), 1203 (C=S), 1391, 1130 (SO₂). ¹H-NMR (DMSO-d₆)δ: 7.2-8.0 [m, 12H, Ar-H+NH], 13.1 [s, 1H, SO₂NH]. MS, m/z (%): 450 [M⁺] (7.0), 344 (100). Anal. Calcd. For C₂₀H₁₄N₆O₃S₂: C, 53.32; H, 3.13; N, 18.66. Found: C, 53.11; H 2.93; N, 18.88.

2. 11 N-benzyl-4-(3-(benzylideneamino)-4-oxo-3,4-dihydroquinazolin-2-ylamino)benzenesulfonamide (14)

A mixture of compound **5** (4.0 g, 0.01 mol) and benzaldehyde (1 g, 0.01 mol) in ethanol (30 mL) was refluxed for 12h. The reaction mixture was filtered while hot and the solid obtained was crystallized from dioxane to give compound **14**. Yield, 72%; m.p. 223-225 °C; IR (KBr, cm⁻¹): 3319, 3203 (NH), 2923, 2845 (CH aliph.), 1665 (C=O), 1394, 1141 (SO₂). ¹H-NMR (DMSO-d₆)δ: 6.9-8.1 [m, 18H, Ar-H+NH], 8.2 (s, 1H, N=CH), 9.8 [s, 1H, SO₂NH]. MS, m/z (%): 496 [M⁺] (0.5), 463 (100). Anal. Calcd. For C₂₆H₂₀N₆O₃S: C, 62.89; H, 4.06; N, 16.39. Found: C, 62.91; H, 4.1; N, 16.42.

2. 12 4-(4-oxo-3-(3-phenylthioureido)-3,4-dihydroquinazolin-2-ylamino)-N-(pyridin-2-yl)benzenesulfonamide (15)

A mixture of **5** (4.0 g, 0.01 mole) and phenyl isothiocyanate (1.35 g, 0.01 mol) in ethanol (30 mL) was refluxed for 5hr. The reaction mixture was filtered while hot and the solid obtained was crystallized from ethanol to give compound **15**. Yield, 51%; m.p.: 76-78 °C. IR (KBr, cm⁻¹): 3240, 3203 (NH), 3036 (CH arom.), 1663 (C=O), 1236 (C=S), 1320, 1100 (SO₂). ¹H-NMR (DMSO-d₆)δ: 4.5

(s, 2H, 2NH of thioureido), 7.1-7.7 [m, 20H, Ar-H+3NH], 11.2 (s, 1H, SO₂NH). MS, m/z (%): 543 [M⁺] (5.01), 135 (100). Anal. Calcd. For C₂₆H₂₁N₇O₃S₂: C, 57.44; H, 3.89; N, 18.04. Found: C, 57.65; H, 3.73; N, 17.88.

2. 13 4-(9-oxo-2-(phenylamino)-[1,2,4]triazolo[5,1-b]quinazolin-3(9H)-yl)-N-(pyridin-2-yl)benzenesulfonamide (16)

A mixture of compound **5** (4.0 g, 0.01 mol) and phenyl isothiocyanate (1.35 g, 0.01 mol) in pyridine (30 mL) was refluxed for 12h. The reaction mixture was poured onto ice/water acidified by dil HCl and the solid obtained was crystallized from dioxane to give compound **16**. Yield, 73%; m.p.: 298-300 °C; IR (KBr, cm⁻¹): 3463, 3234 (NH), 3045 (CH arom.), 1663 (C=O), 1385, 1139 (SO₂). ¹H-NMR (DMSO-d₆)δ: 7.2-8.1 [m, 18H, Ar-H+NH], 13.1 (s, 1H, SO₂NH). MS, m/z (%): 509 [M⁺] (15.6), 57 (100). Anal. Calcd. For C₂₆H₁₉N₇O₃S: C, 61.29; H, 3.76; N, 19.24. Found: C, 61.44; H, 3.92; N, 19.50.

3. Results and Discussion

3.1. Synthesis

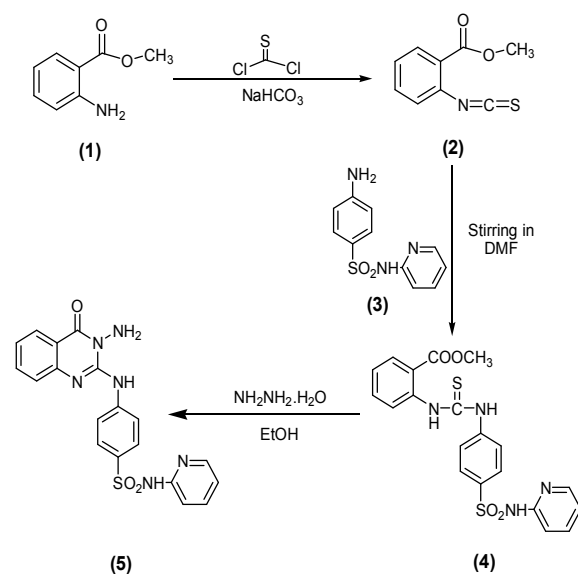
The starting material methyl-2-isothiocyanatobenzoic acid **2** was synthesized from the reaction of methylanthranilate with thiophosgene [43]. The reactivity of isothiocyanato derivative **2** towards nitrogen nucleophile was investigated. When compound **2** was reacted with sulfapyridine **3** in dimethylformamide at room temperature, the thioureido derivative **4** was obtained in good yield (Scheme 1). The structure of compound **4** was supported by elemental analysis, IR, ¹H-NMR and mass spectral data. IR spectrum of compound **4** showed the absence of NCS band and the presence of bands at 3300, 3238 cm⁻¹ (NH), 1651 cm⁻¹ (C=O), 1262 cm⁻¹ (C=S), 1389, 1133 cm⁻¹ (SO₂), 1262 cm⁻¹ (C=S). The ¹H-NMR spectrum of compound **4** in (DMSO-d₆) revealed signal at 3.6 ppm due to (OCH₃) group. The mass spectrum of compound **4** revealed a molecular ion peak m/z at 429 [M⁺-CH₃] (0.4%), with a base peak m/z at 226 (100%).

Treatment of compound **4** with hydrazine hydrate in ethanol afforded the corresponding N-amino derivative **5**. The formation of compound **5** was assumed to occur via elimination of 1 mol H₂S followed by intramolecular cyclization [44] to give the N-amino derivative **5** (lead acetate paper). The IR spectrum of compound **5** showed bands at 3410, 3308, 3183 cm⁻¹ (NH, NH₂), 1660 cm⁻¹ (C=O), 1383, 1133 cm⁻¹ (SO₂). The ¹H-NMR spectrum of compound **5** in (DMSO-d₆) revealed signals at 6.8 ppm assigned to (NH₂) group, 13.1 ppm corresponding to (SO₂NH) group. The mass spectrum of compound **5** revealed a molecular ion peak m/z at

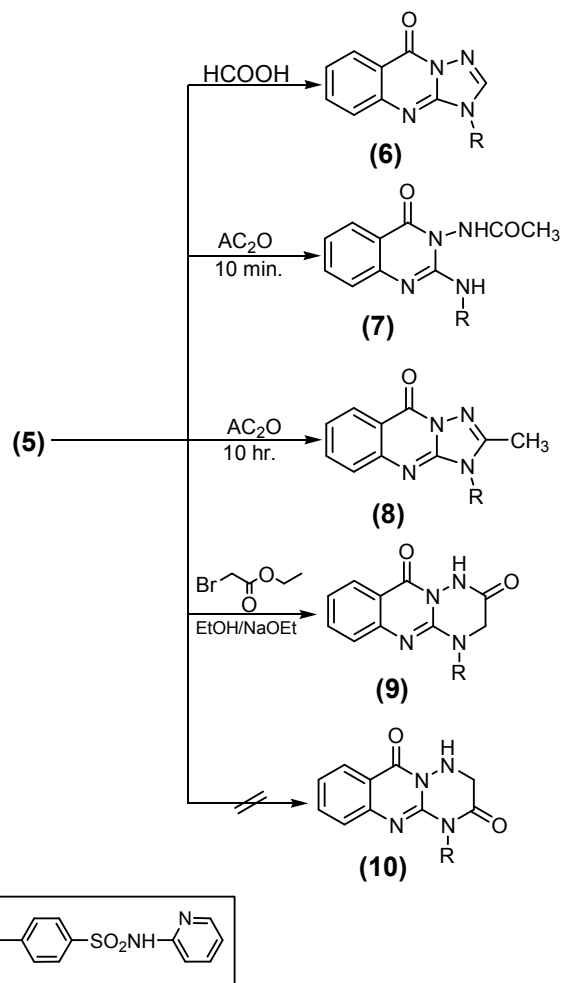
408 [M⁺] (2.1%), with a base peak m/z at 343 (100%).

The reactivity of N-amino derivative **5** was studied. Thus, interaction of compound **5** with formic acid yielded the 1,2,4-triazolo[5,1-*b*]quinazoline derivative **6** (Scheme 2). The structure of compound **6** was established by elemental analysis, IR, ¹H-NMR and mass spectroscopy. The IR spectrum showed bands at 3227 cm⁻¹(NH), 1663 cm⁻¹ (C=O), 1395, 1139 cm⁻¹ (SO₂). The ¹H-NMR spectrum of compound **6** in (DMSO-d₆) revealed signals at 8.0 ppm assigned to CH of triazole ring and a multiplet at 6.9-7.8 ppm for aromatic protons. The mass spectrum of compound **6** revealed a molecular ion peak m/z at 418 [M⁺] (1.65%), with a base peak m/z at 88 (100%). When compound **5** was refluxed with acetic anhydride for short time (10 min.), the corresponding mono acetyl derivative **7** was obtained. While, for long time (10 hr) gave the cyclic compound triazoloquinazoline **8** (Scheme 2). The structure of compounds **7** and **8** was established by elemental analysis, IR and ¹H-NMR spectral data. The IR spectrum of compound **7** showed bands at 3190 cm⁻¹ (NH), 1721, 1662 cm⁻¹ (2 C=O), 1393, 1142 cm⁻¹ (SO₂). The ¹H-NMR spectrum of compound **7** in (DMSO-d₆) revealed signals at 2.6 ppm for (COCH₃) group, 11.7 ppm due to (NH) group. The IR spectrum of compound **8** showed bands at 3272 cm⁻¹ (NH), 1686 cm⁻¹(C=O), 1359, 1169 cm⁻¹ (SO₂). Its ¹H-NMR spectrum in (DMSO-d₆) revealed signals at 1.2 ppm assigned to (CH₃) group. Reaction of **5** with ethyl bromoacetate in refluxing sodium ethoxide yielded the triazinoquinazoline **9** rather than its isomeric structure **10** (Scheme 2). Structure **9** was suggested rather than structure **10**, based on the assumption that the reaction basic condition allowed it to proceed through formation of sodium salt on the less basic NH, and elimination of sodium bromide followed by cyclization [45]. The IR spectrum of the isolated compound showed bands at 3484 cm⁻¹ (NH), 1743, 1675 cm⁻¹ (2 C=O) which was at less frequency than that expected for structure **10**. Further evidence was the ¹H-NMR spectrum which showed a singlet at 3.9 ppm for the methylene protons. When compound **5** was reacted with ethyl cyanoacetate under condition of fusion the 3-cyanoacetamido-quinazoline derivative **11** was obtained in good yield, while in the presence of triethylamine as catalyst, the cyclic triazoloquinazoline **12** was obtained (Scheme 3). The structure of compound **11** and **12** was established by

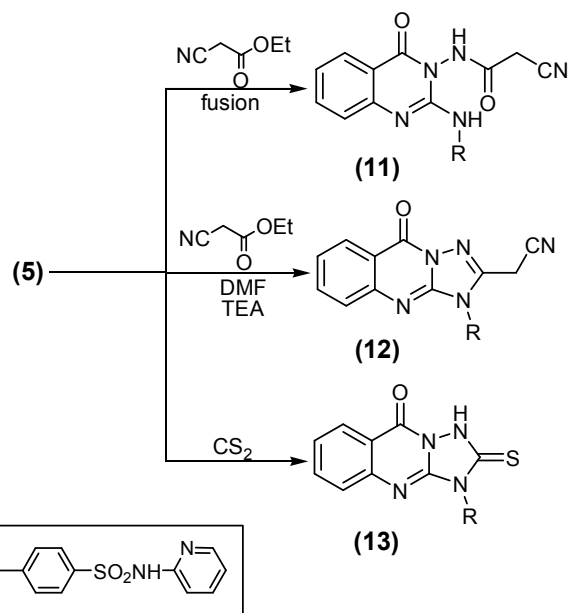
elemental analysis, IR, ¹H-NMR, mass spectral data. IR spectrum of compound **11** showed bands at 3223 cm⁻¹ (NH), 2208 cm⁻¹ (C≡N), 1652 cm⁻¹(C=O), 1398, 1139 cm⁻¹ (SO₂). The ¹H-NMR spectrum of compound **11** in (DMSO-d₆) exhibited signals at 4.1 ppm for (CH₂) group and 13.1 ppm due to (SO₂NH) group. The mass spectrum of compound **11** revealed a molecular ion peak m/z at 475 [M⁺] (0.15%), with a base peak m/z at 373 (100%). The IR spectrum of compound **12** showed bands at 3356 cm⁻¹ (NH), 2206 cm⁻¹ (C≡N), 1645 cm⁻¹(C=O), 1393, 1139 cm⁻¹ (SO₂). The ¹H-NMR spectrum of compound **12** in (DMSO-d₆) revealed signal at 4.2 ppm corresponding to (CH₂) group. The mass spectrum of compound **12** revealed a molecular ion peak m/z at 457 [M⁺] (18.8%), with a base peak m/z at 92 (100%). On the other hand, refluxing compound **5** with carbon disulfide in pyridine yielded the 2-thioxo-1,2,4-triazolo[5,1-*b*]quinazoline derivative **13**. Its IR spectrum showed bands at 3223 cm⁻¹ (NH), 1663 cm⁻¹(C=O), 1391, 1130 cm⁻¹ (SO₂), 1250 cm⁻¹(C=S). The ¹H-NMR spectrum of compound **13** in (DMSO-d₆) revealed signal at 13.1 due to (SO₂NH) group. The mass spectrum of compound **13** revealed a molecular ion peak m/z at 450 [M⁺] (7.0%), with a base peak m/z at 344 (100%).



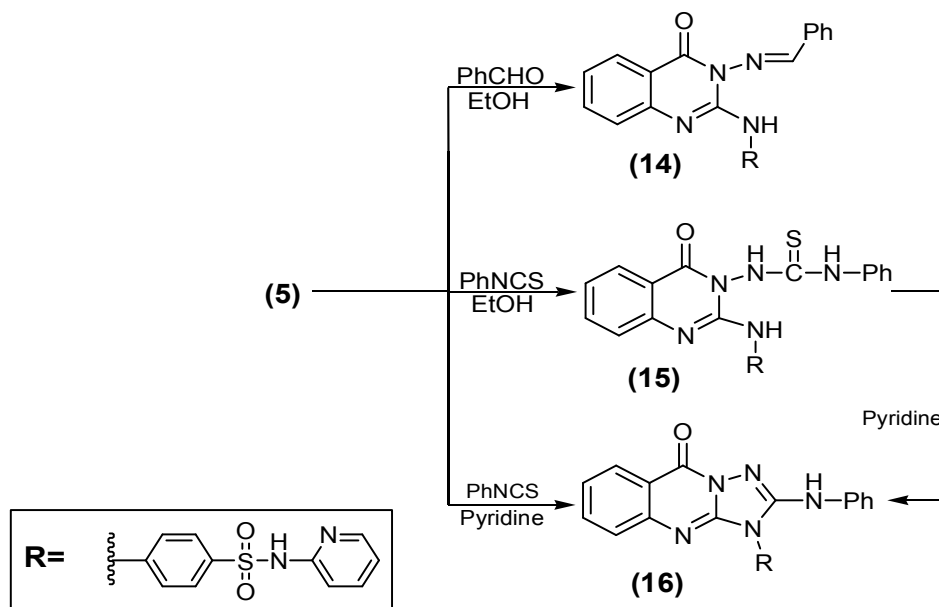
Scheme 1. Synthesis of novel thioureido and N-aminoquinazoline **4,5**



Scheme 2. Synthesis of quinazoline, triazolo and triazinoquinolines **6-9**.



Scheme 3. Synthesis of quinazoline and triazoloquinazolines **11-13**.



Scheme 4. Synthesis of quinazolines and triazoloquinazoline **14-16**.

The Schiff base **14** was obtained via reaction of compound **5** with benzaldehyde in refluxing ethanol (Scheme 4). The structure of compound **14** was confirmed by elemental analysis, IR, $^1\text{H-NMR}$, mass spectral data. The IR spectrum of compound **14** showed bands at 3319, 3203 cm^{-1} (NH), 1665 cm^{-1} (C=O), 1394, 1141 cm^{-1} (SO₂). Its $^1\text{H-NMR}$ spectrum in (DMSO- d_6) revealed signal at 8.2 ppm due to (N=CH). The mass spectrum of compound **14** revealed a molecular ion peak m/z at 496 [M^+] (0.5%), with a base peak m/z at 463 (100%). Our work was extended to study the reactivity of compound **5** towards phenyl isothiocyanate. Thus, reaction of **5** with phenyl isothiocyanate in refluxing ethanol furnished the corresponding thioureido derivative **15**, while in refluxing pyridine, the triazoloquinazoline **16** was obtained. The structure of compounds **15** and **16** was established by elemental analyses, IR, $^1\text{H-NMR}$ and mass spectral data. The IR spectrum of compound **15** showed bands at 3240, 3203 cm^{-1} (NH). The $^1\text{H-NMR}$ spectrum of compound **15** in (DMSO- d_6) revealed signals at 4.5 ppm assigned to (2NH of thioureido group). The IR spectrum of compound **16** showed bands at 3463, 3234 cm^{-1} (NH), 1663 cm^{-1} (C=O), 1356, 1139 cm^{-1} (SO₂). The mass spectrum of compound **16** revealed a molecular ion peak m/z at 509 [M^+](15.6%), with a base peak m/z at 57(100%). Also, compound **16** was obtained by refluxing compound **15** in pyridine (m.p and mixed m.p).

3.1. In vitro anticancer screening

Human tumor liver cell line (HEPG2) was used in this study. The cytotoxic activity was measured *in vitro* for the newly synthesized compounds using the Sulfo-Rhodamine-B stain (SRB) assay using the method of Skehan et al. [46]. The *in vitro* anticancer screening was done by the pharmacology unit at the National Cancer Institute, Cairo University. Cells were plated in 96-multiwell plate (10^4 cells/well) for 24 hours before treatment with the compound(s) to allow attachment of cell to the wall of the plate. Test compounds were dissolved in dimethyl sulfoxide. Different concentrations of the compound under test (5, 12.5, 25, and 50 μM) were added to the cell monolayer. Triplicate wells were prepared for each individual concentration. Monolayer cells were incubated with the compound(s) for 48 h at 37 °C and in atmosphere of 5% CO₂. After 48 h, cells were fixed, washed and stained for 30 min with 0.4% (wt/vol) SRB dissolved in 1% acetic acid. Excess unbound dye was removed by four washes with 1% acetic acid and attached stain was recovered with Tris-EDTA buffer. Color intensity was measured in an ELISA reader. The relation between surviving fraction and drug concentration is plotted to get the survival curve for liver tumor (HEPG2) cell line after the specified time. The molar concentration required for 50% inhibition of cell viability (IC₅₀) was calculated and compared to the reference drug doxorubicin (CAS, 25316-40-9). The surviving fractions were expressed as means \pm standard error and the results are given in Table 1.

Table 1. In-vitro cytotoxic activity of the synthesized compounds against human liver cell line (HEPG2)

Comp. No.	Compound Concentration (μM)				IC ₅₀
	5	12.5	25	50	
	Surviving Fraction (Means \pm SE) [#]				
Dox.	0.09673 \pm 0.02470	0.06157 \pm 0.01168	0.06333 \pm 0.01011	0.1011 \pm 0.04738	4.15
4	0.9047 \pm 0.03370	0.4781 \pm 0.0007424	0.3114 \pm 0.01646	0.3497 \pm 0.00647	26.90
5	0.7773 \pm 0.06803	0.6979 \pm 0.03952	0.3901 \pm 0.01669	0.3536 \pm 0.01818	30.32
6	0.9456 \pm 0.01762	0.5438 \pm 0.05800	0.4047 \pm 0.03188	0.3822 \pm 0.01740	30.84
7	0.8744 \pm 0.008724	0.5541 \pm 0.03543	0.4843 \pm 0.03407	0.4997 \pm 0.01106	38.42
8	0.8567 \pm 0.06306	0.8783 \pm 0.03450	0.5998 \pm 0.03279	0.6051 \pm 0.02456	55.32
9	1.011 \pm 0.01620	0.9473 \pm 0.03460	0.9102 \pm 0.06981	0.7637 \pm 0.06663	105.63
11	0.8220 \pm 0.05584	0.7150 \pm 0.04493	0.3544 \pm 0.01235	0.3255 \pm 0.04075	29.16
12	0.7573 \pm 0.02083	0.6388 \pm 0.06557	0.4477 \pm 0.02468	0.4675 \pm 0.02854	35.67
13	0.8379 \pm 0.01204	0.7931 \pm 0.08121	0.4440 \pm 0.03875	0.4596 \pm 0.02073	37.63
14	0.2313 \pm 0.01331	0.4242 \pm 0.09302	0.4886 \pm 0.09658	0.4404 \pm 0.1207	22.11
15	0.4880 \pm 0.01783	0.3878 \pm 0.01496	0.3588 \pm 0.02462	0.3182 \pm 0.01561	19.70
16	0.8643 \pm 0.01502	0.6162 \pm 0.04373	0.2911 \pm 0.005154	0.3634 \pm 0.02558	28.30

[#]: Each value is the mean of three values \pm Standard Error

3.2 Radiosensitizing evaluation

The rationale for combining chemotherapy and radiotherapy is based mainly on two ideas, one being spatial cooperation, which is effective if chemotherapy is sufficiently active to eradicate subclinical metastases and if the primary local tumor is effectively treated by radiotherapy. In this regard, no interaction between radiotherapy and chemotherapy is required. The other idea is the enhancement of radiation effects by direct enhancement of the initial radiation damage by incorporating drugs into DNA, inhibiting cellular repair, accumulating cells in a radiosensitive phase or eliminating radioresistant phase cells, eliminating hypoxic cells, or inhibiting the accelerated repopulation of tumor cells. Virtually, all chemotherapeutic agents have the ability to sensitize

cancer cells to the lethal effects of ionizing radiation (Nishimura, 2004).

The most potent compounds resulted from the *in vitro* anticancer screening; the quinazalone derivatives **14** and **15**, were selected to be evaluated again for their *in vitro* anticancer activity alone and in combination with γ -radiation. This study was conducted to evaluate the ability of these compounds to enhance the cell killing effect of γ -radiation. Cells were subjected to a single dose of γ -radiation at a dose level of 8 Gy with a dose rate of 2 Gy/min. Irradiation was performed in the National Cancer Institute, Cairo University, using Gamma cell-40 (⁶⁰Co) source. The surviving fractions were expressed as means \pm standard error. The results were analyzed using 1-way ANOVA test and given in Table 2.

Table (2): *In vitro* anticancer screening of compounds **14** and **15** against human liver cell line (HEPG2) in combination with γ -radiation.

Compd. No.	Control	Irradiated (8 Gy)	Compound Concentration (μM) + Irradiation (8 Gy)				IC ₅₀ (μM)
			5	12.5	25	50	
			Surviving Fraction (Means \pm SE) [#]				
Dox.	1.000	0.927 \pm 0.02*	0.113 \pm 0.0012*	0.4266 \pm 0.04702*	0.53 \pm 0.08505*	0.96 \pm 0.111*	3.67
14	1.000	0.927 \pm 0.02*	0.498 \pm 0.0712*	0.3111 \pm 0.01396*	0.3221 \pm 0.01176*	0.3442 \pm 0.02365*	17.92
15	1.000	0.927 \pm 0.02*	0.4766 \pm 0.03177*	0.2911 \pm 0.03588*	0.3001 \pm 0.00329*	0.2992 \pm 0.04411*	15.71

[#]: Each value is the mean of three values \pm Standard Error

*: Significant difference from control group at p<0.001

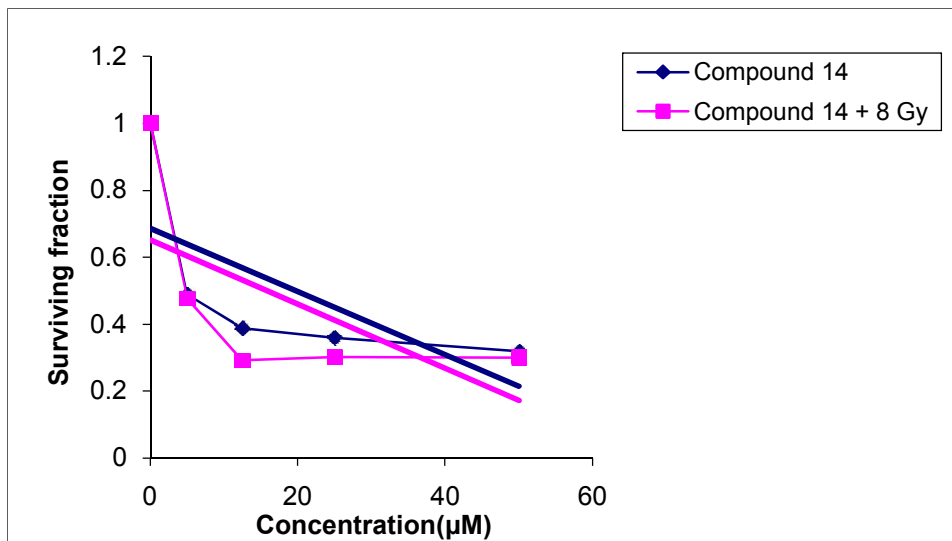


Figure (1): Survival curve for HEPG2 cell line for compound 14 alone or in combination with γ -irradiation (8 Gy)

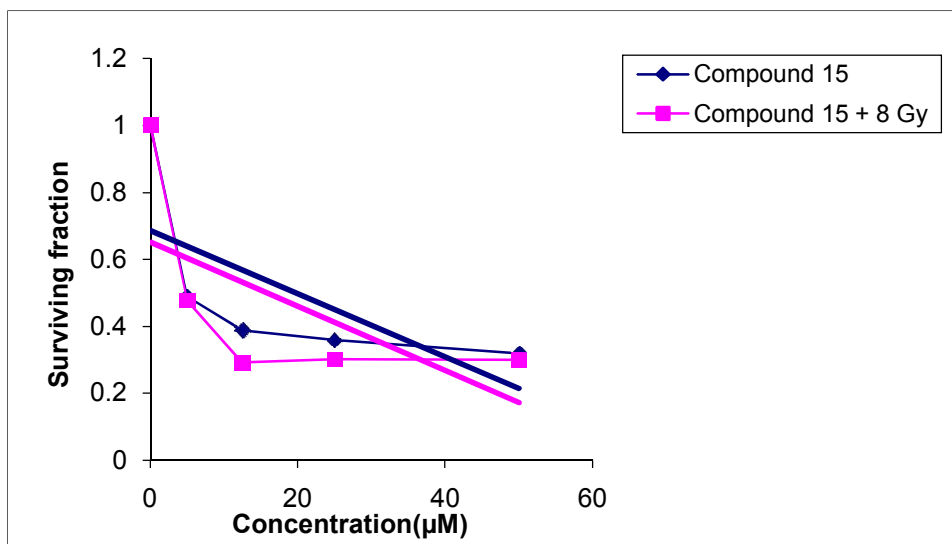


Figure (2): Survival curve for HEPG2 cell line for compound 15 alone or in combination with γ -irradiation (8 Gy)

4. Conclusion

From the above results, we can conclude that administration of the tested compounds **14** and **15** on human liver (HEPG2) cell line showed promising cytotoxic activity with IC_{50} value of **22.11** and **19.70** μM , respectively. While, combining these compounds with radiation at the same concentrations resulted in a remarkable improvement of their activity with IC_{50} value of **17.92** and **15.71** μM , respectively. This demonstrates the importance of the combination therapy (CT and RT) for the patients with cancer to decrease the side effects of both drugs and radiation.

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Corresponding author

Mostafa M. Ghorab^{1,2}

¹Department of Pharmacognosy, College of Pharmacy, King Saud University, P.O. Box 2457, Riyadh 11451, Saudi Arabia.

²Department of Drug Radiation Research, National Centre for Radiation Research & Technology, Atomic Energy Authority, Cairo, Egypt.

mmsghorab@yahoo.com

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